

- The enthalpy of neutralisation is the heat energy released when 1 mole of water is formed from neutralisation of an acid with a base.
- Neutralisation is an exothermic reaction. The reaction between NaOH with HCl can be written as:



- By omitting the spectator ions the reaction becomes:



Exothermic Reaction Neutralisation Results

Oswaal Editorial Board



Exothermic Reaction Neutralisation Results:

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