Name

Sharanya Saxena

Date:

16 October, 2020

Student Exploration: Ionic Bonds

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: chemical family, electron affinity, ion, ionic bond, metal, nonmetal, octet rule, shell, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

 Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

Since there are 8 markers in a set, Nate would have to give one marker to Clara:

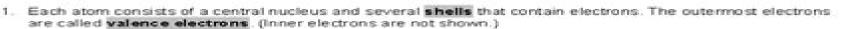
Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

Maggie can give up two markers, one of which can go to each Fred and Florence.

Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The *lonic Bonds* Gizmo allows you to explore how **lonic bonds** form.

To begin, check that **Sodium (Na)** and **Chlorine (Cl)** are selected from the menus at right. Click **Play** (•) to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



How many valence electrons does each atom have? Sodium: 1 Chlorine: 7

- - A. Try pulling an electron away from each atom. Based on this experiment, which atom is a metal?

Sodium Which is a nonmetal? Chlorine

B. Try moving an electron from the metal to the nonmetal. What happens?

Sodium lets go of its one electron.

Exploration Guide Covalent Bonds Answer Key

David L. Nelson, Albert L. Lehninger, Michael M. Cox

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